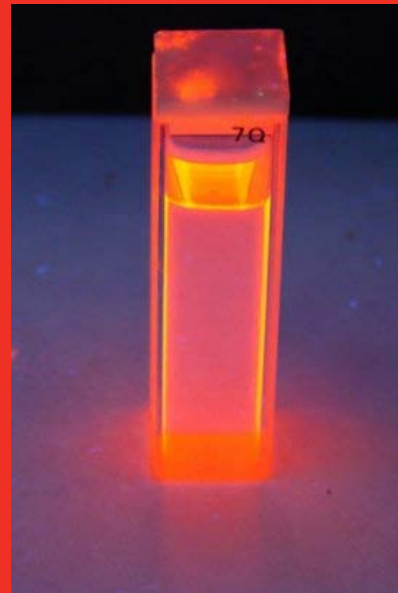
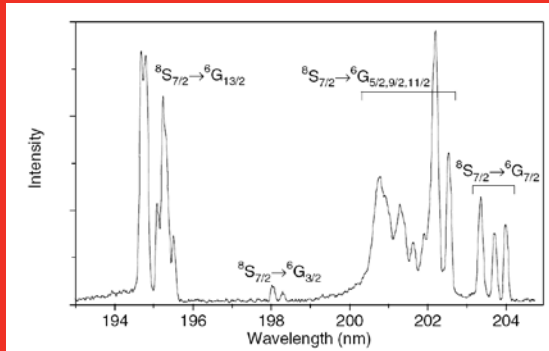


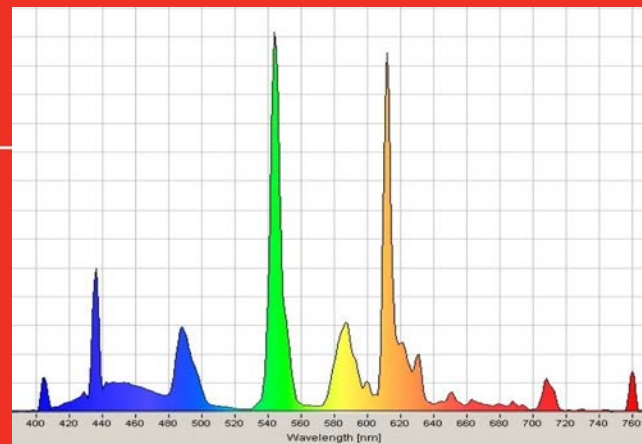
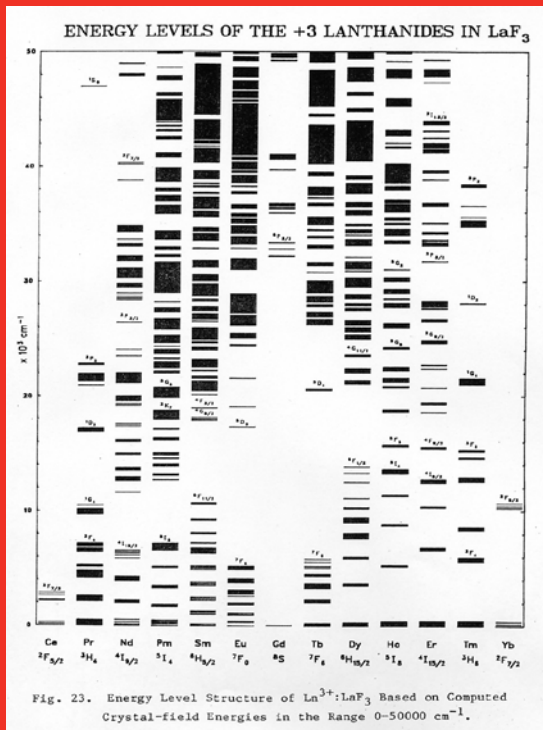
Electronic Structure and Transition Intensities in Rare-Earth Materials



Michael F Reid

School of Physical and Chemical Sciences
Te Kura Matū
 University of Canterbury
 Christchurch, New Zealand

<http://www.phys.canterbury.ac.nz>

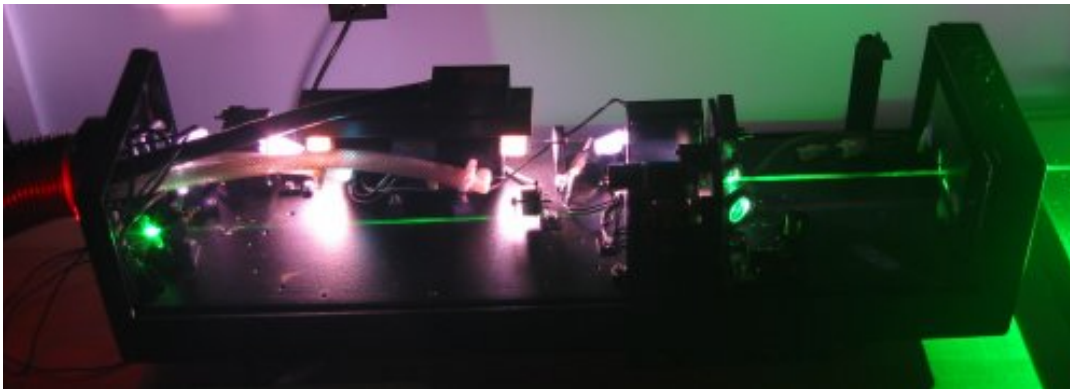


January 2018 v01



Outline

- Background on rare-earth (lanthanide) ions.
- States and transitions
- Effective Hamiltonian for $4f^N$
- Some simple calculations of energy levels
- Transition intensities
- $4f^{N-1}5d$
- Ab-initio calculations



Gerhard Dieke

Johns Hopkins 1950-1965



Bill Carnall



1960's Theory: Judd, Ofelt, Wybourne



Filling of orbitals

S

IA	IIA	IIIB																			
1 H 1.008																					
3 Li 6.939	4 Be 9.0122																				
11 Na 22.99	12 Mg 24.312																				
19 K 39.102	20 Ca 40.08	21 Sc 44.956																			
37 Rb 85.468	38 Sr 87.62	39 Y 88.906																			
55 Cs 132.91	56 Ba 137.33	57 La 138.91	58 Ce 140.12	59 Pr 140.91	60 Nd 144.24	61 Pm *145	62 Sm 150.36	63 Eu 151.96	64 Gd 157.25	65 Tb 158.93	66 Dy 162.51	67 Ho 164.93	68 Er 167.26	69 Tm 168.93	70 Yb 173.04	71 Lu 174.97					
87 Fr *223	88 Ra 226.03	89 Ac 227.03	90 Th 232.04	91 Pa 231.04	92 U 238.03	93 Np 237.05	94 Pu *244	95 Am *243	96 Cm *247	97 Bk *247	98 Cf *251	99 Es *252	100 Fm *257	101 Md *258	102 No *259	103 Lr *260					

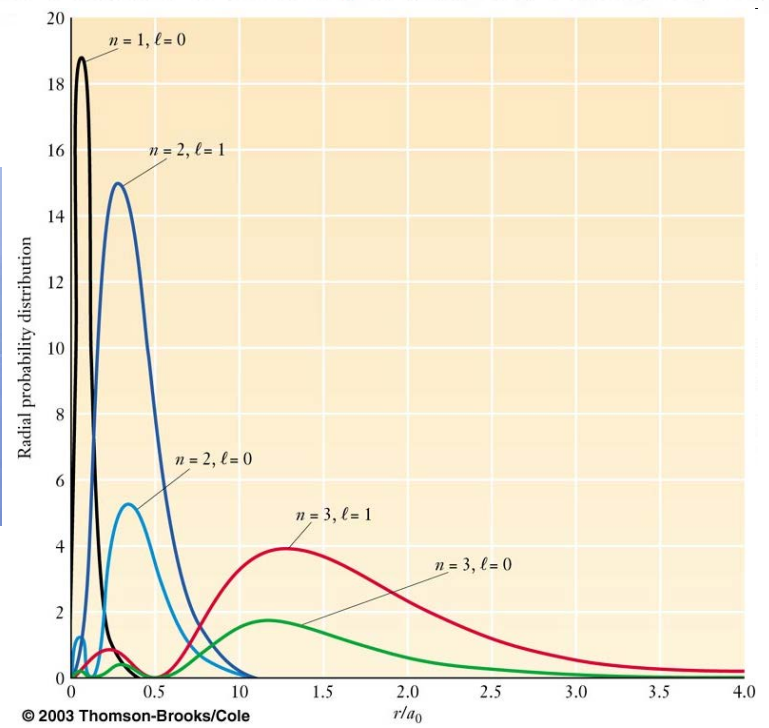
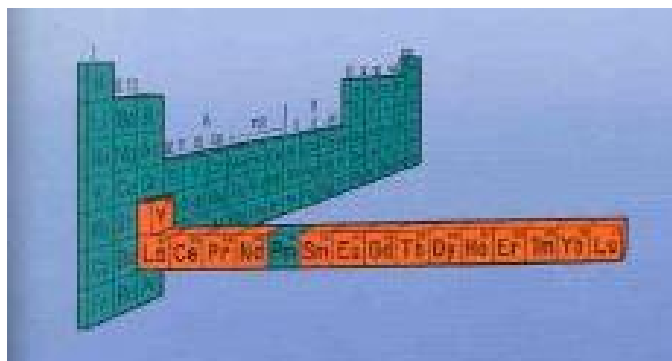
f

IVB	VB	VIB	VII B	VIII B	IX B	X B	IB	II B	III	IVA	VA	VI A	VII A	VIIIA	VIIIA
									1 H 1.008					1 H 1.008	2 He 4.0026
									5 B 10.811	6 C 12.011	7 N 14.007	8 O 15.999	9 F 18.998	10 Ne 20.183	
									13 Al 26.982	14 Si 28.086	15 P 30.974	16 S 32.064	17 Cl 35.453	18 Ar 39.948	
22 Ti 47.88	23 V 50.942	24 Cr 51.996	25 Mn 54.938	26 Fe 55.847	27 Co 58.932	28 Ni 58.71	29 Cu 63.54	30 Zn 65.37	31 Ga 69.72	32 Ge 72.59	33 As 74.922	34 Se 78.96	35 Br 79.909	36 Kr 83.8	
40 Zr 91.224	41 Nb 92.906	42 Mo 95.94	43 Tc *96	44 Ru 101.07	45 Rh 102.91	46 Pd 106.42	47 Ag 107.9	48 Cd 112.41	49 In 114.82	50 Sn 118.71	51 Sb 121.75	52 Te 127.61	53 I 126.9	54 Xe 131.29	
72 Hf 178.49	73 Ta 180.95	74 W 183.85	75 Re 186.21	76 Os 190.2	77 Ir 192.22	78 Pt 195.08	79 Au 196.97	80 Hg 200.29	81 Tl 204.38	82 Pb 207.2	83 Bi 208.98	84 Po *209	85 At *210	86 Rn *222	
104 Rf 101.07	105 Ha 102.91	106 Sg 106.42	107 Ns 107.87	108 Hs 108.91	109 Mt 109.90	110 Uun 110.96	111 Uuu 111.96	112 Uub 112.96	113 Uuq *285	114 Uuh *289	115 Uuq *289	116 Uuh *289	117 Uuq *293	118 Uuo *293	

d

s

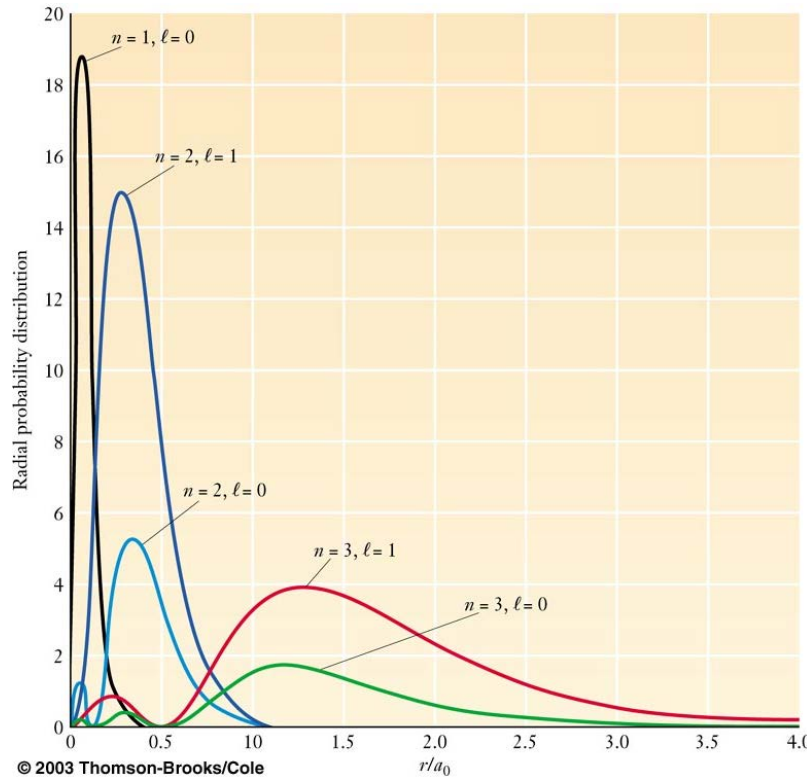
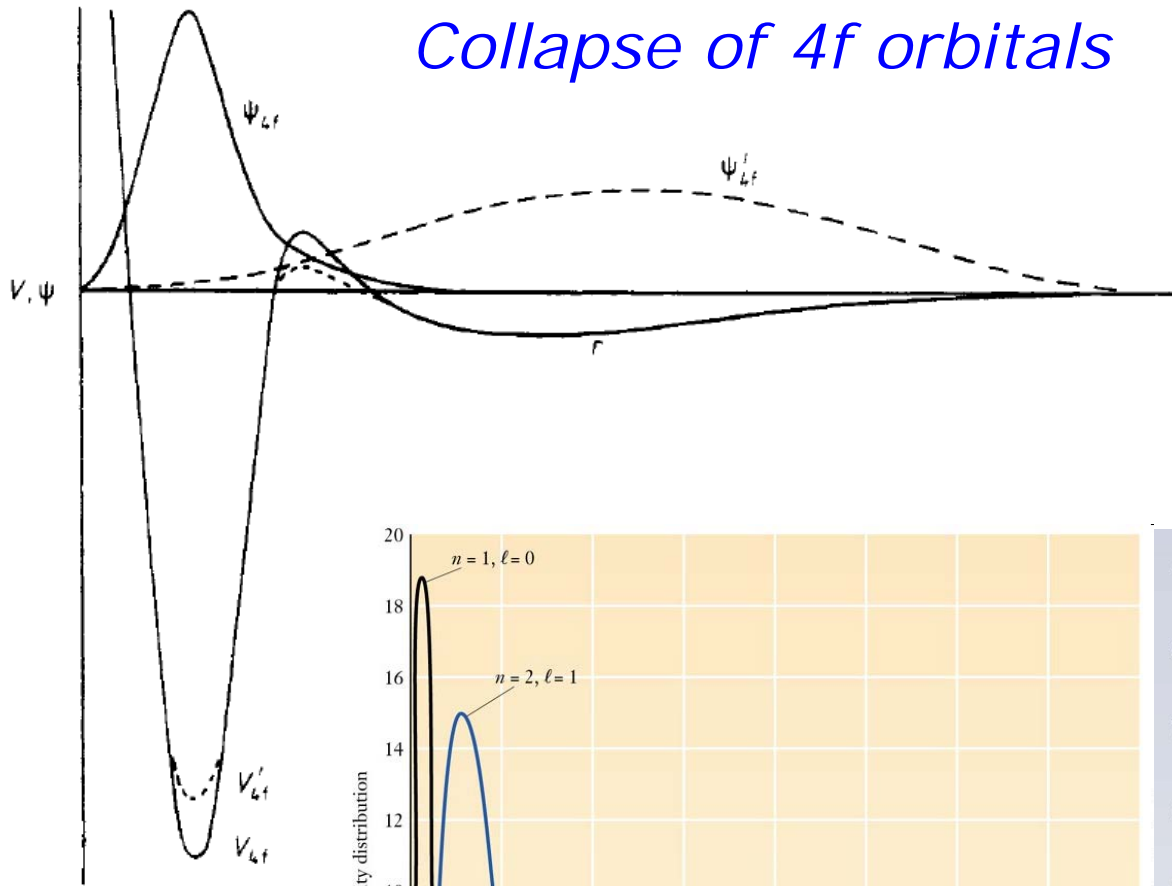
Based on symbols used by ACS S.M. Andrews, 2001



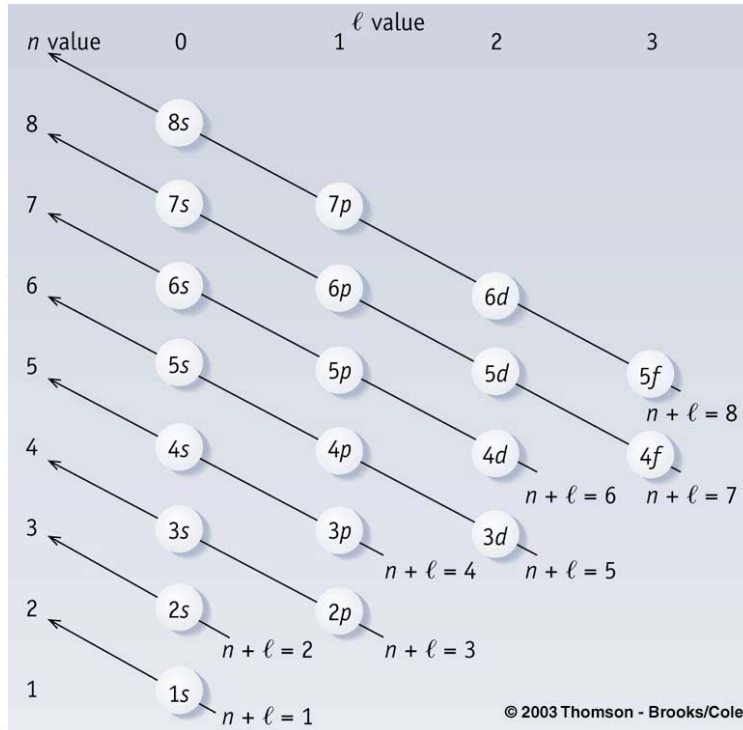
© 2003 Thomson-Brooks/Cole

Collapse of 4f orbitals

JP Connerade, 1982

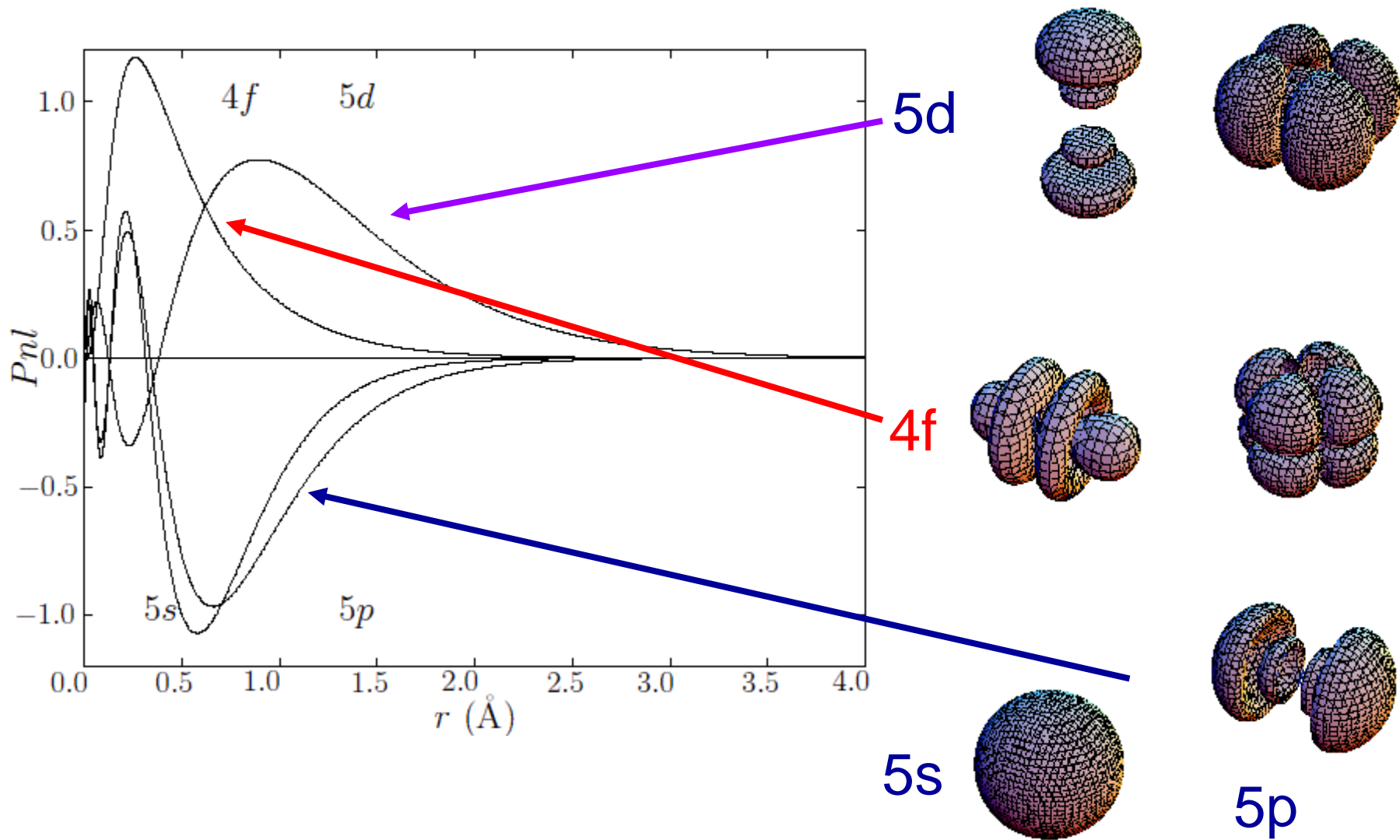


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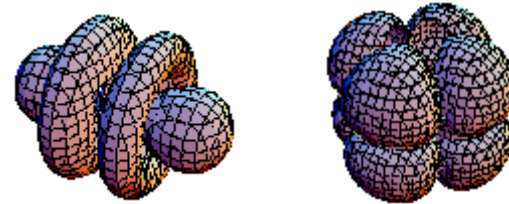
Lanthanide 3+ ground state: $5s^2 5p^6 4f^N 5d^0$



Lanthanides: $4f^N$, $4f^{N-1}5d$, Excitons

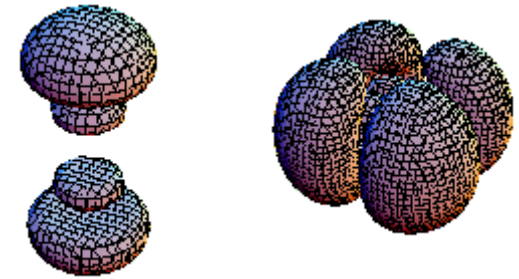
▪ $4f^N$

- Sharp lines
- Long lifetimes
- So ideal for laser and phosphor applications



▪ $4f^{N-1}5d$

- Broad absorption bands from $4f^N$
- Useful for absorbing energy
- Short lifetimes useful in some applications, such as scintillators

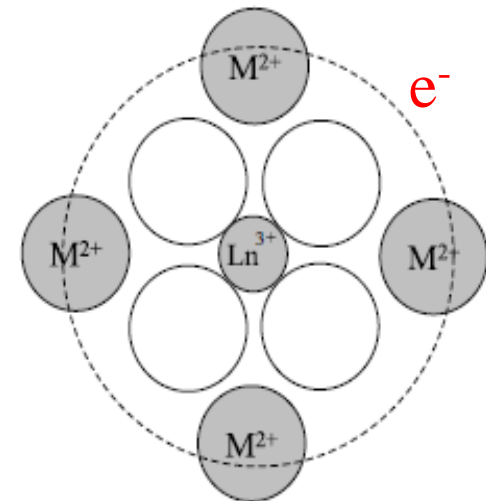


▪ Excitons

- Excited electron can become delocalized, giving an excitonic state.

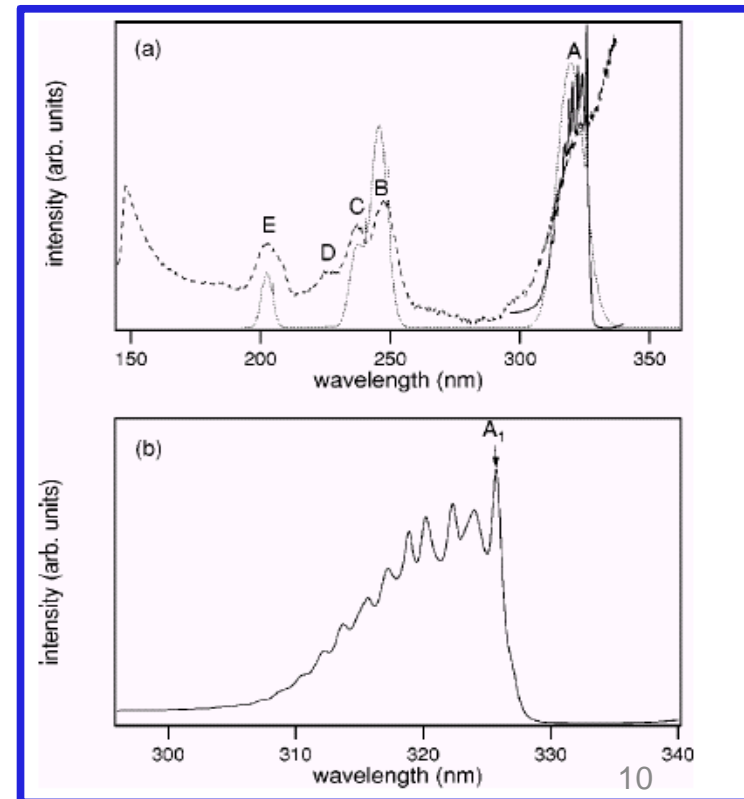
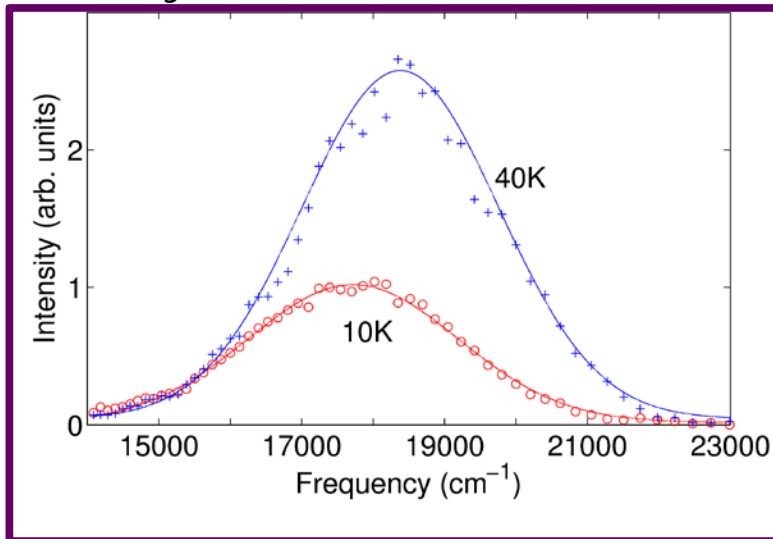
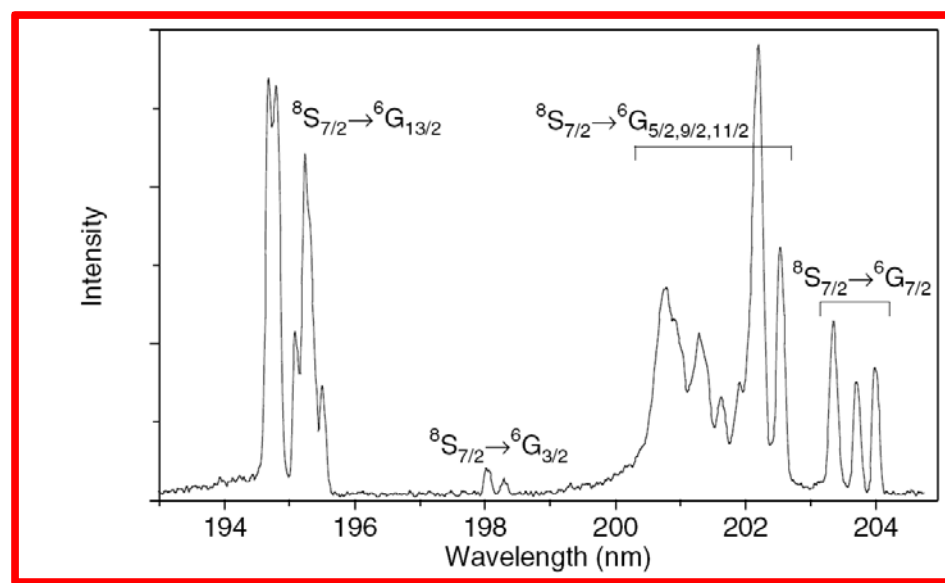
▪ Charge Transfer Transitions

- Ligand to lanthanide electron transfer



Transitions

- $4f^N - 4f^N$
 - No configuration shift
 - Sharp lines
- $4f^N - 4f^N 5d$
 - Configuration shift
 - Broad bands
- Excitonic states $\rightarrow 4f^N$
 - Large configuration shift
 - Very broad bands



How do we proceed?

- Ab-initio (first principles) calculations
 - Well established in atoms
 - Hartree-Fock + perturbations
 - Now viable for lanthanide complexes but slow
- Effective Hamiltonians (“crystal field”)
 - Requires parameter fitting
 - Relatively quick and easy, allowing rapid interpretation of spectra.
 - Can be related to ab-initio calculations.

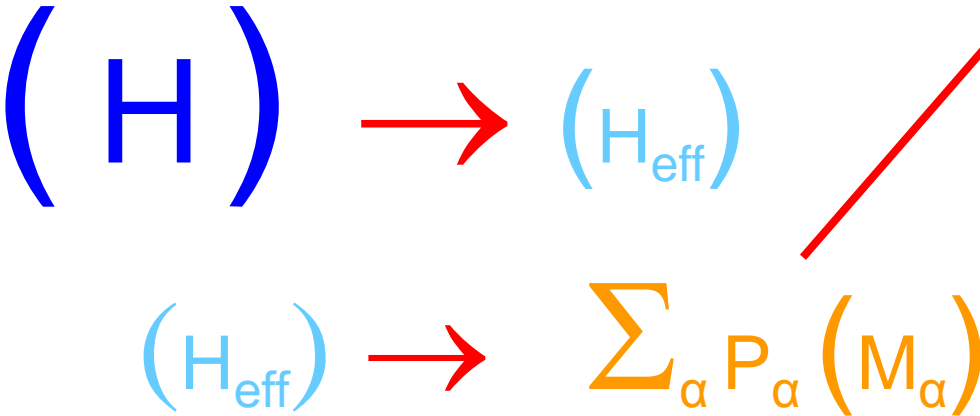
Effective Hamiltonian Calculations

- $H |\Psi_i\rangle = E_i |\Psi_i\rangle$ (H is Hamiltonian) =

- $H_{\text{eff}} |\phi_i\rangle = E_i |\phi_i\rangle$ (H_{eff} is Effective Hamiltonian)

=

=

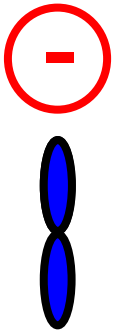
 E_{Theory} E_{Expt}


z

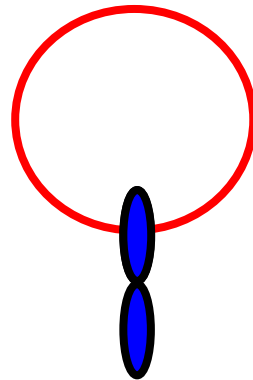


“Crystal Field”

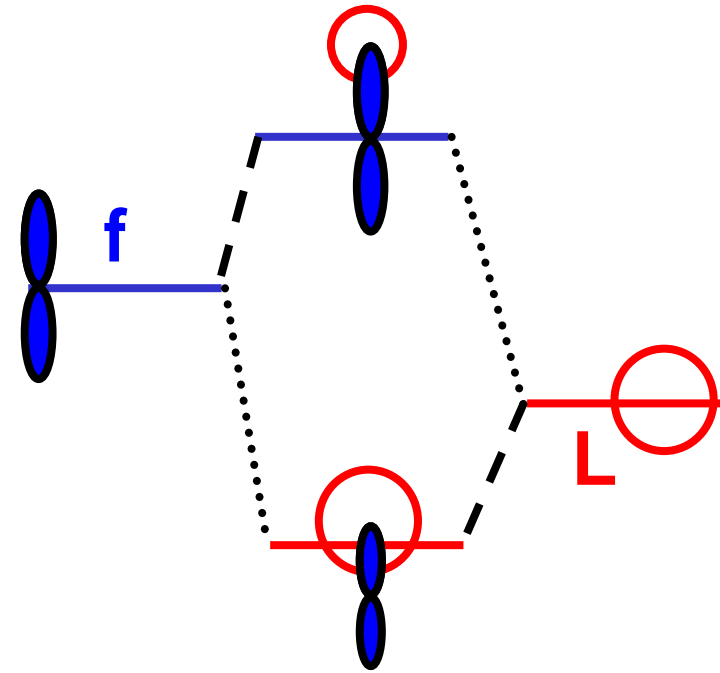
$$\left(\sum_{k,q} B_q^k C_q^{(k)} \right)$$



Electrostatic



Overlap

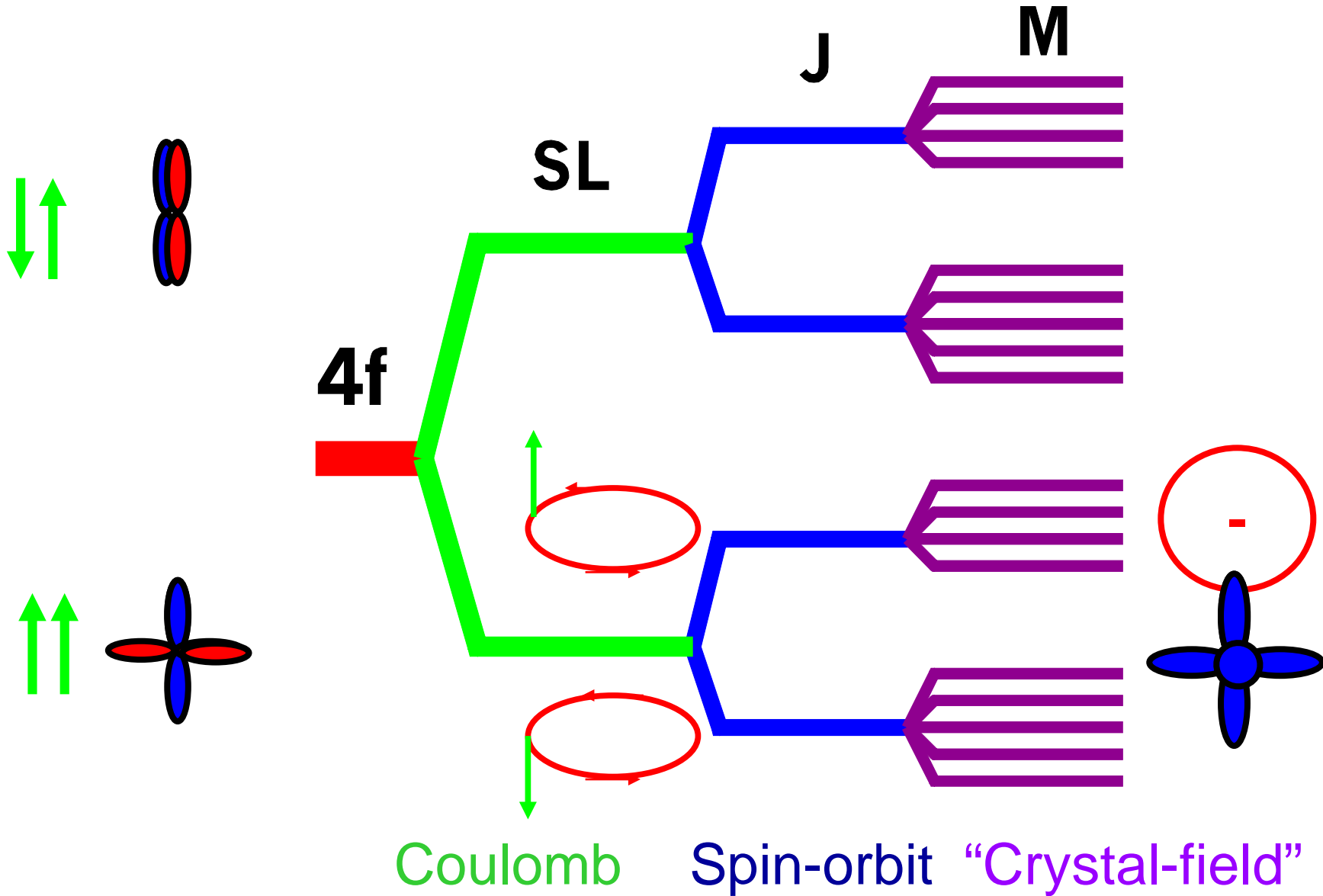


Covalency

All increase energy of z orbital more than x,y

Orbital energies ↔ crystal-field parameters

Understanding the energy levels: $4f^N$



Effective Hamiltonian for $4f^N$

$$H = E_{avg} + \sum_{k=2,4,6} F^k f_k + \zeta_f A_{so} + \sum_{k,q} B_q^k C_q^{(k)}$$

Coulomb
Spin-Orbit
Crystal Field

Correlation and other corrections

$$+ \alpha L(L+1) + \beta G(G_2) + \gamma G(R_7) + \sum_{i=2,3,4,6,7,8} T^i t_i$$

$$+ \sum_{h=0,2,4} M^h m_h + \sum_{k=2,4,6} P^k p_k$$

Calculating Matrix elements

- Wigner-Eckart Theorem

$$\langle \alpha J M | T_q^{(k)} | \alpha' J' M' \rangle = (-1)^{J-M} \begin{pmatrix} J & k & J' \\ -M & q & M' \end{pmatrix} \langle \alpha J || T^{(k)} || \alpha' J' \rangle$$

matrix element “geometrical” factors “reduced”
matrix element

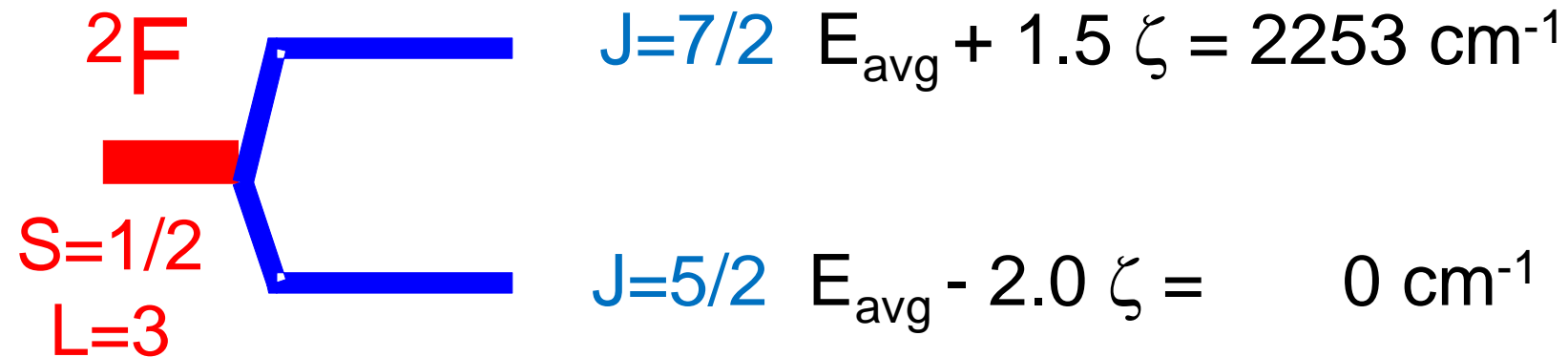
- Selection Rules

- $M' + q = M$
- $J \ k \ J'$ form a triangle: $|J - J'| \leq k \leq J + J'$

- $|SLJM\rangle$ states

- More complex versions of WET.
- Triangle rules for $S \ S' \ L \ L' \ J \ J'$ and operator labels.

Ce³⁺: 4f¹ 5d¹



Free-ion splitting is 2253 cm^{-1} so $\zeta = 644 \text{ cm}^{-1}$

Similarly for $5d^1$

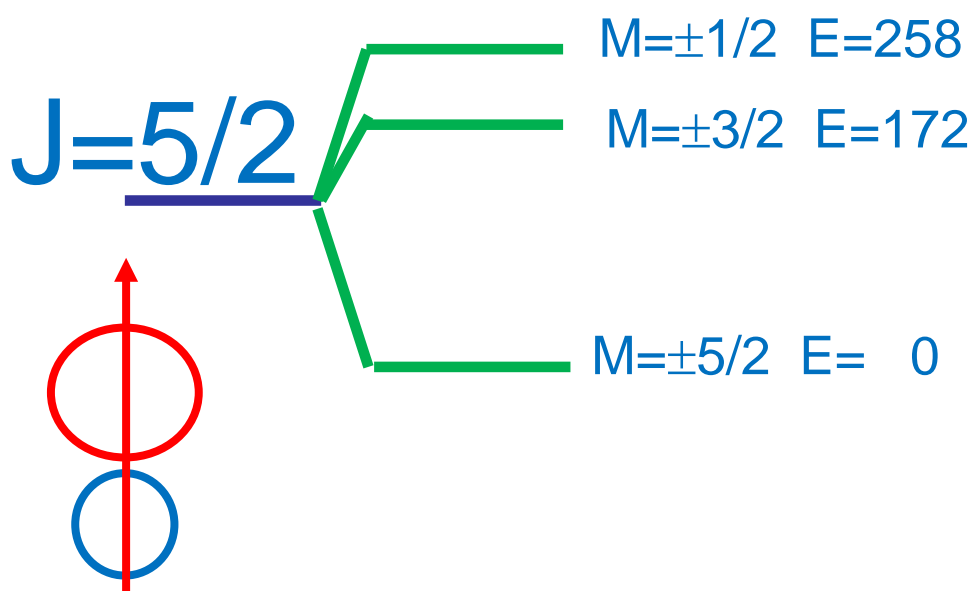
Splitting of $J=5/2$ multiplet



- **Magnetic field**

- Even spacing:

- 0.4 cm^{-1} for 1T field



- **Crystal Field**

$$H_{\text{cf}} = \sum_{k,q} B_q^k C_q^{(k)}$$

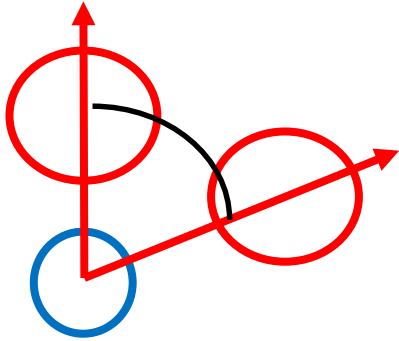
- Uneven spacing

- Example:

- $B_0^2 = 500 \text{ cm}^{-1}$

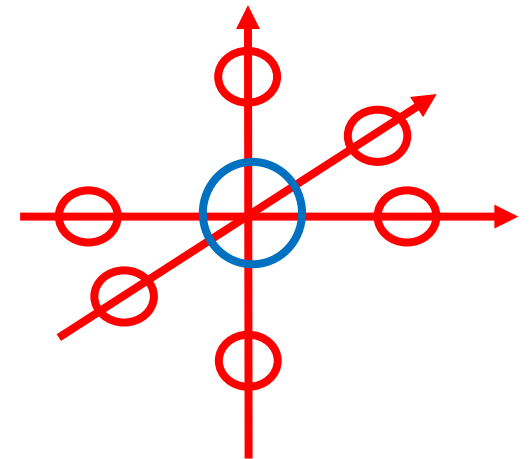
- More complex cases mix up M labels.

Superposition Model



$$B_q^k = \sum_L \bar{B}_k(R_0) (-1)^q C_{-q}^{(k)}(\theta_L, \phi_L) \left(\frac{R_0}{R_L} \right)^{t_k}$$

$$C_0^{(2)}(x, y, z) = \frac{1}{r^2} \sqrt{\frac{1}{4}} (3z^2 - r^2)$$

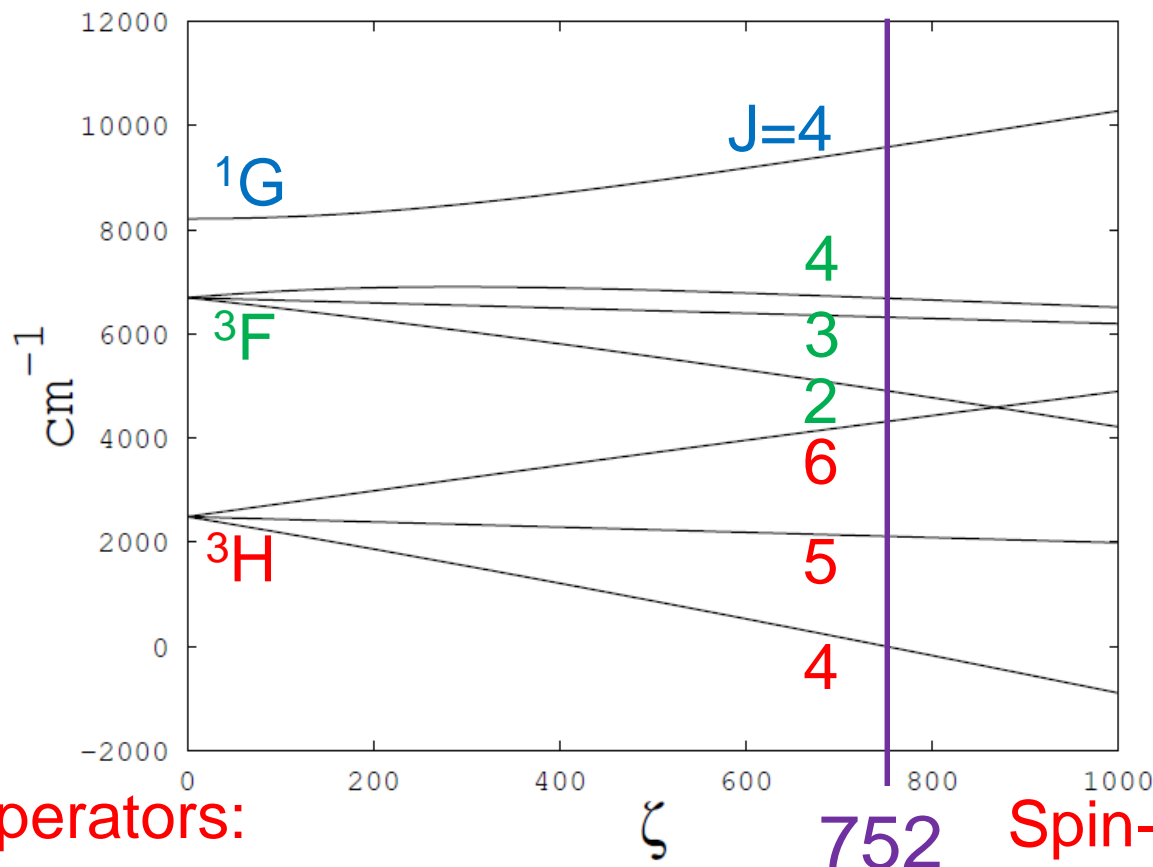


Pr³⁺ : 4f² Coulomb + Spin-orbit

$$E(^3H_J) = -0.0906F^2 - 0.0328F^4 + 0.0162F^6 = -7359 \text{ cm}^{-1},$$

$$E(^3F_J) = -0.0239F^2 - 0.0163F^4 - 0.0209F^6 = -3155 \text{ cm}^{-1},$$

$$E(^1G_4) = -0.1128F^2 + 0.1031F^4 + 0.0285F^6 = -1641 \text{ cm}^{-1}.$$

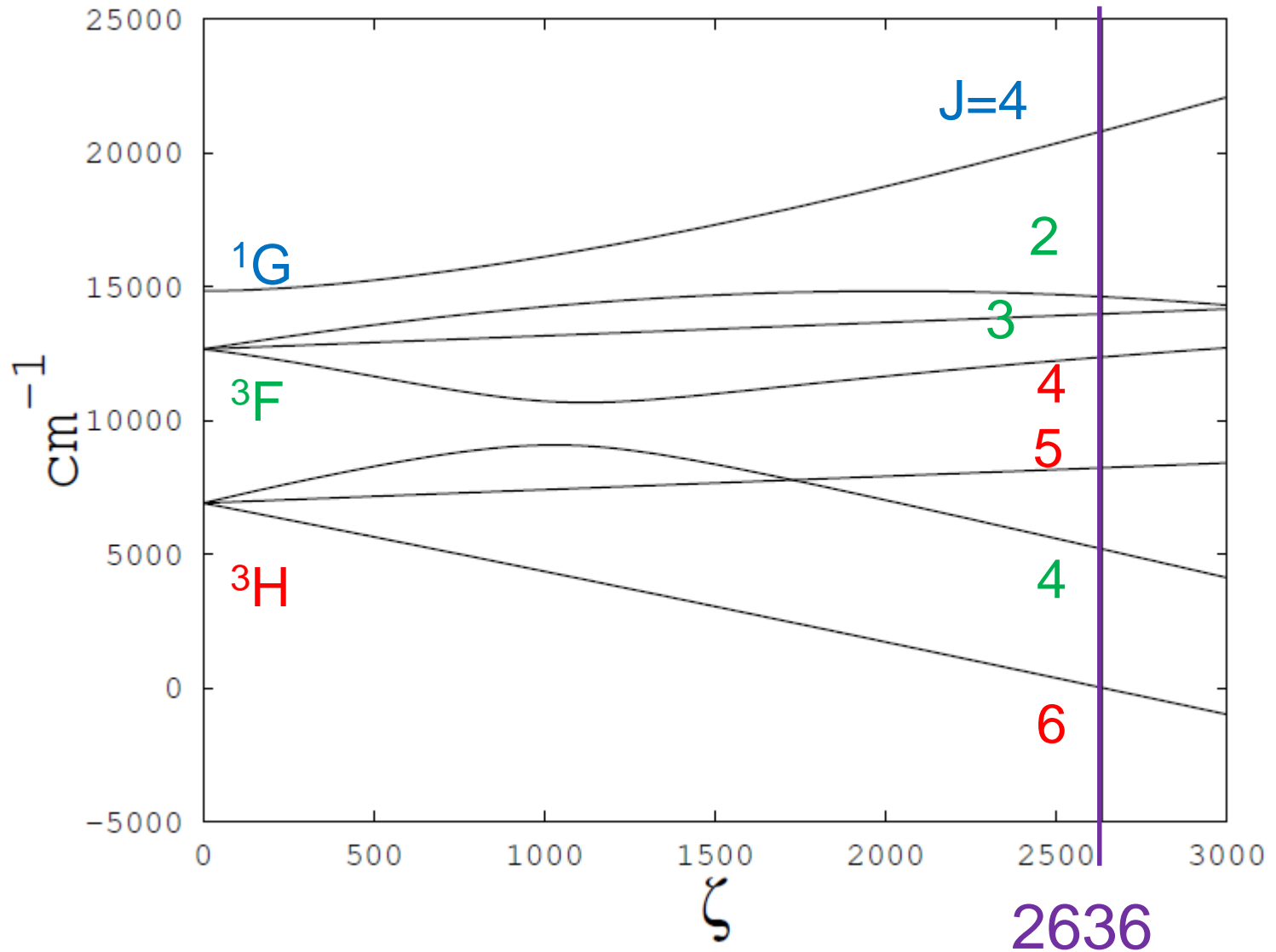


Coulomb operators:
S=0, L=0, J=0

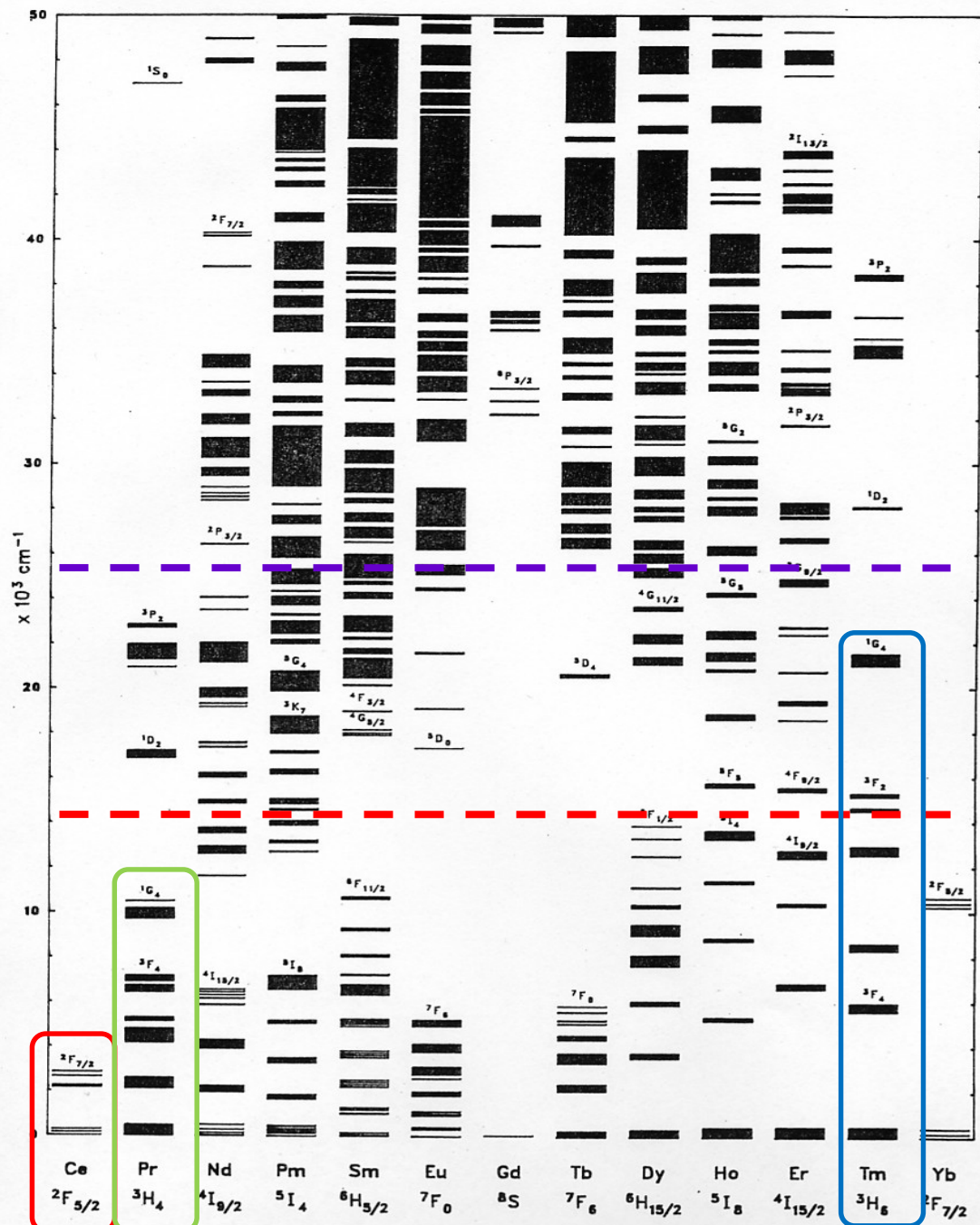
Spin-orbit operator:
S=1, L=1, J_z=0

Tm³⁺: 4f¹²

Spin-orbit much larger and matrix elements change sign.



ENERGY LEVELS OF THE +3 LANTHANIDES IN LaF_3



Transition Intensities

- Electric Dipole, Magnetic Dipole, ...
- ED between $4f^N$ and $4f^{N-1}5d$ can be calculated directly
 - But require modelling of vibronic bands.
- ED within $4f^N$ are parity forbidden.
 - Construct Effective ED operator that accounts for mixing of configurations of opposite parity on ion or ligand.
 - First detailed treatment: Judd, Ofelt, 1962.

Effective Electric Dipole Operator

$$D_{\text{eff},q} = D_q^{(1)} + D_q^{(1)} \sum_{\beta \notin M} \frac{|\beta\rangle\langle\beta|V}{E_0 - E_\beta^{(0)}} + \sum_{\beta \notin M} \frac{V|\beta\rangle\langle\beta|}{E_0 - E_\beta^{(0)}} D_q^{(1)} + \dots$$

Can derive a parametrization. $\lambda=2,4,6, t=\lambda\pm 1, \lambda$

$$D_{\text{eff},q} = \sum_{\lambda,t,p} A_{tp}^\lambda U_{p+q}^{(\lambda)} (-1)^q \langle \lambda(p+q), 1-q | tp \rangle$$

Dipole strength

$$S_{FI,q}^{\text{ED}} = \sum_i \sum_f e^2 \left| \langle Ff | D_q^{(1)} | Ii \rangle \right|^2$$

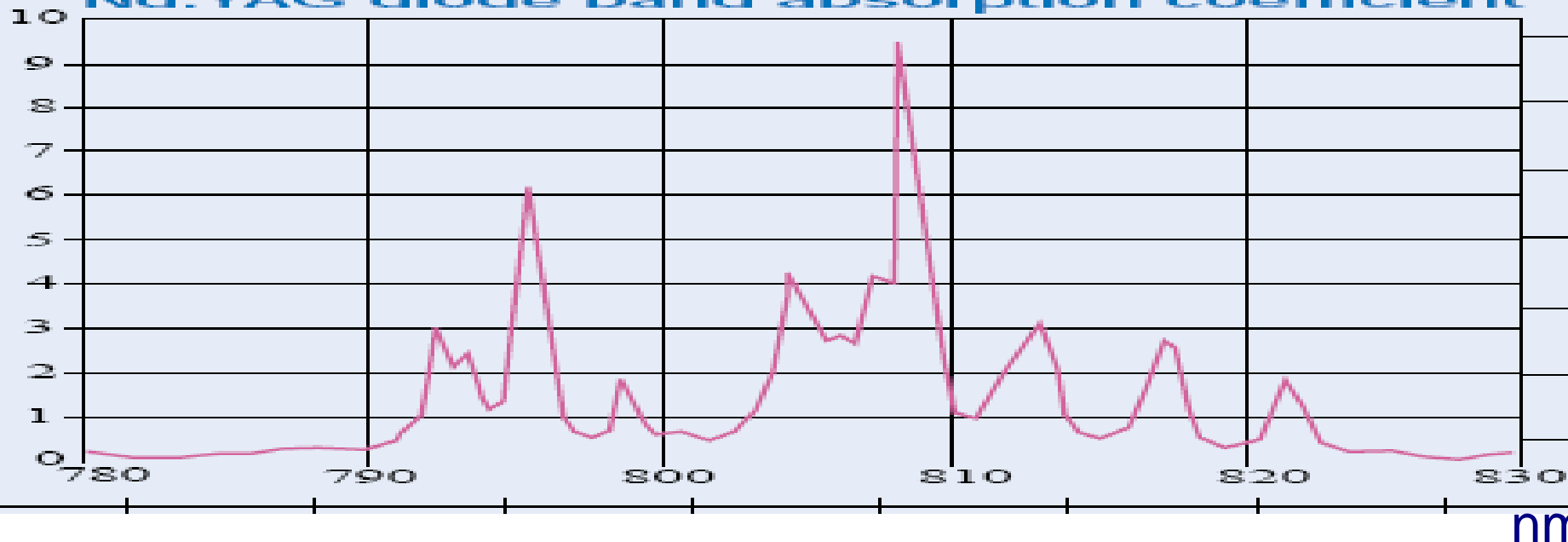
Oscillator strength

$$f_{FI,q}^{\text{ED}} = \frac{2m\omega}{\hbar e^2} \frac{\chi_L}{n} \frac{1}{g_I} S_{FI,q}^{\text{ED}}$$

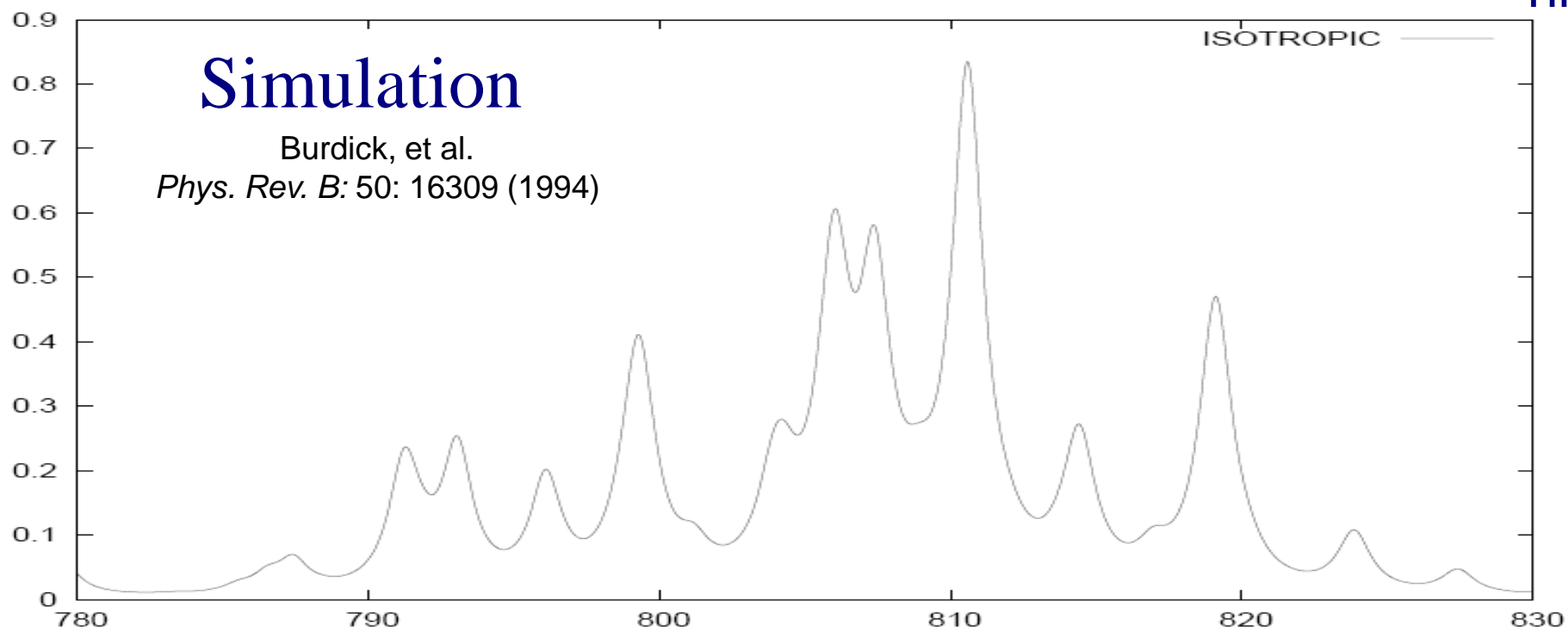
Einstein A coefficients ($1/\tau$)

$$A_{FI,q}^{\text{ED}} = \frac{1}{4\pi\epsilon_0} \frac{4\omega^3}{\hbar c^3} n \chi_L \frac{1}{g_I} S_{FI,q}^{\text{ED}}$$

Nd:YAG diode band absorption coefficient



nm

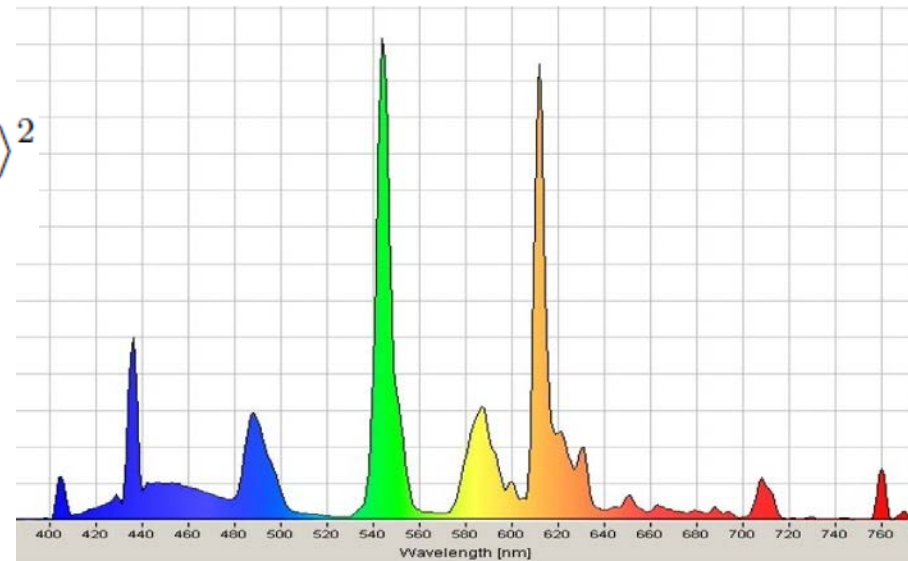


Multiplet-Multiplet transitions

- **Judd 1962**
 - For solutions and glasses at room temperature.
 - Sum over all states in a multiplet and all polarizations.
 - Reduces to three-parameter *linear* fit.
 - Ω_λ parameters with $\lambda=2,4,6$
 - **Over 3000 citations!**

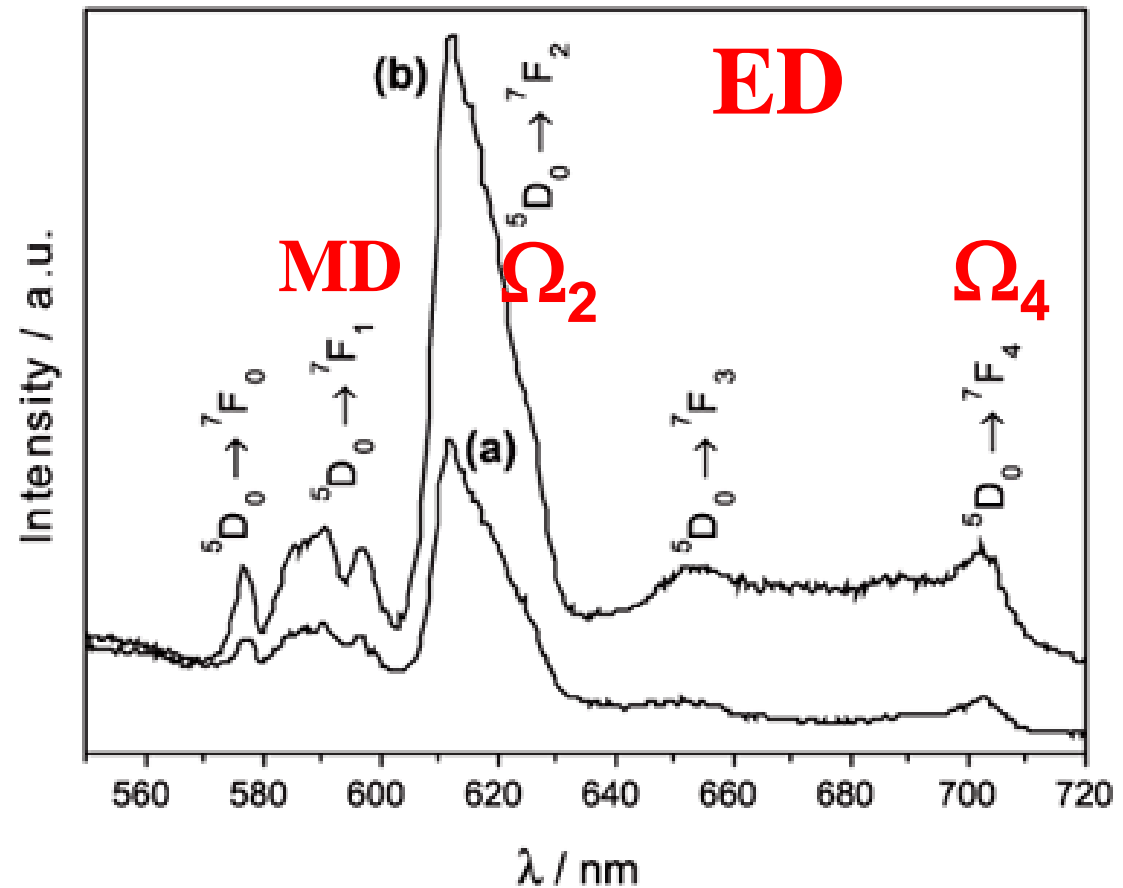
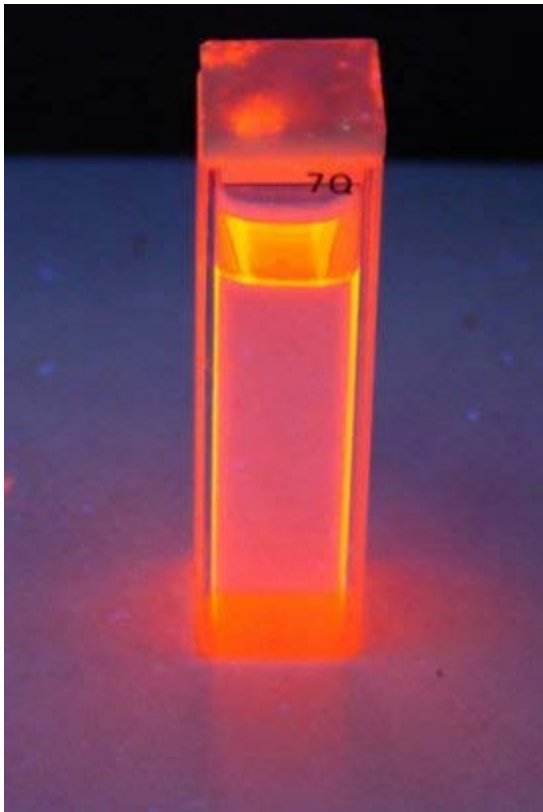
$$\bar{S}_{\alpha_F J_F, \alpha_I J_I}^{\text{ED}} = \frac{1}{3} e^2 \sum_{\lambda} \Omega_{\lambda} \langle \alpha_F J_F \| \mathbf{U}^{(\lambda)} \| \alpha_I J_I \rangle^2$$

$$\Omega_{\lambda} = \sum_{t,p} \frac{1}{2\lambda + 1} |A_{tp}^{\lambda}|^2$$



Eu³⁺: $^5D_0 \rightarrow ^7F_J$ Emission

$$\frac{1}{3}e^2 \sum_{\lambda} \Omega_{\lambda} \langle \alpha_F J_F \| U^{(\lambda)} \| \alpha_I J_I \rangle^2$$



Eliminate options: Pr-Yb downconversion

Linda Aarts ,
Spectroscopy Letters, 43:373–381, 2010

$$\frac{1}{3}e^2 \sum_{\lambda} \Omega_{\lambda} \langle \alpha_F J_F \| \mathbf{U}^{(\lambda)} \| \alpha_I J_I \rangle^2$$

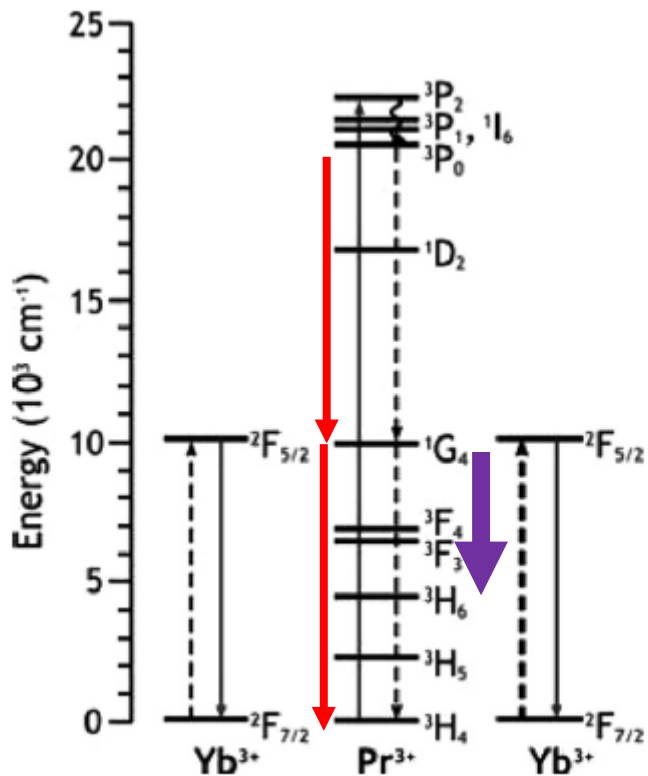


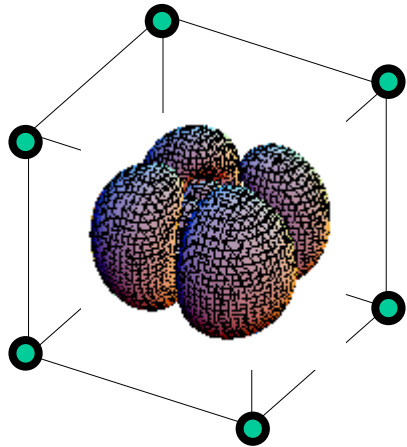
TABLE 1 Reduced Matrix Elements $(U^{(t)})^2$ for Transition Starting from the $\text{Pr}^{3+}1G_4$ Level

Transition	$(U^{(2)})^2$	$(U^{(4)})^2$	$(U^{(6)})^2$
$1G_4 \rightarrow 3H_4$	0.00141	0.00635	0.02206
$1G_4 \rightarrow 3H_5$	0.03739	0.09615	0.41314
$1G_4 \rightarrow 3H_6$	0.25226	0.25337	0.23683
$1G_4 \rightarrow 3F_2$	0.00004	0.01580	0.00587
$1G_4 \rightarrow 3F_3$	0.00381	0.00531	0.05173
$1G_4 \rightarrow 3F_4$	0.07819	0.14271	0.34419
Sum	0.37310	0.51968	1.07382
Percentage ($1G_4 \rightarrow 3H_4$)	0.38	1.2	2.1

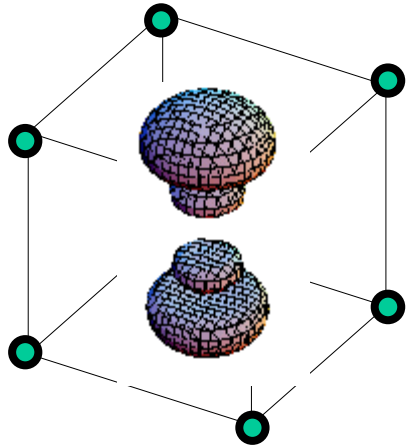
The percentages in the bottom row give the $(U^{(t)})^2$ strength for the $1G_4 \rightarrow 3H_4$ transitions relative to the sum of all $(U^{(t)})^2$ values for a given value of t .

$1G_4 \rightarrow 3H_4$ can never be strong...

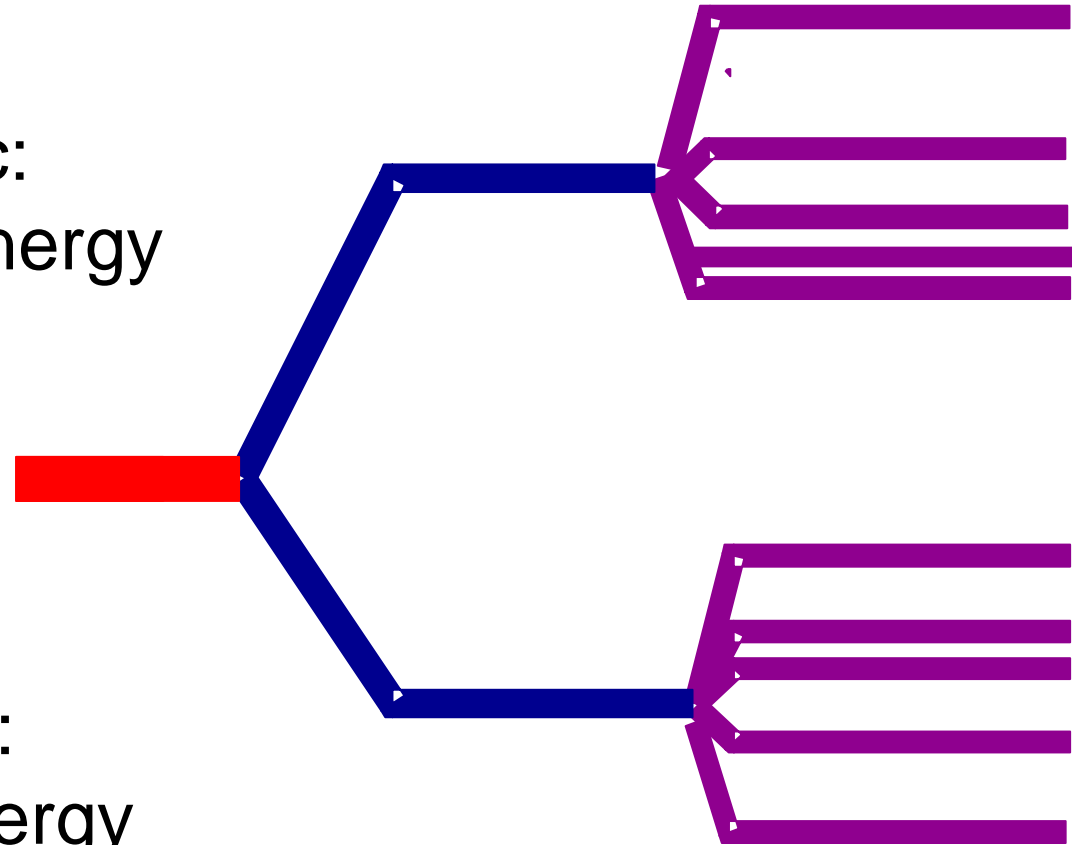
Understanding the energy levels: $4f^{N-1}5d$



T_2
Cubic:
higher energy



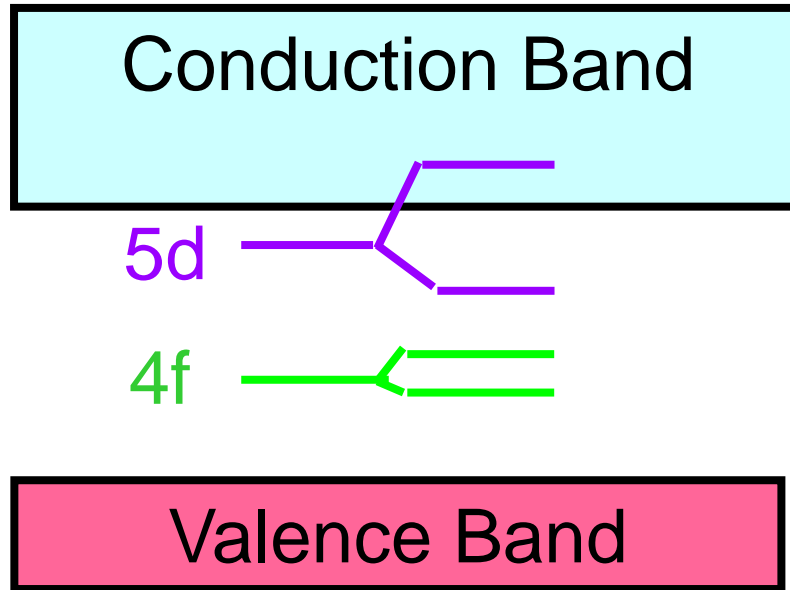
E
Cubic:
lower energy



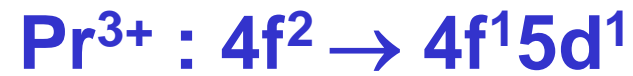
Crystal field Coulomb, etc

40 x 4f (20,000 vs 500)

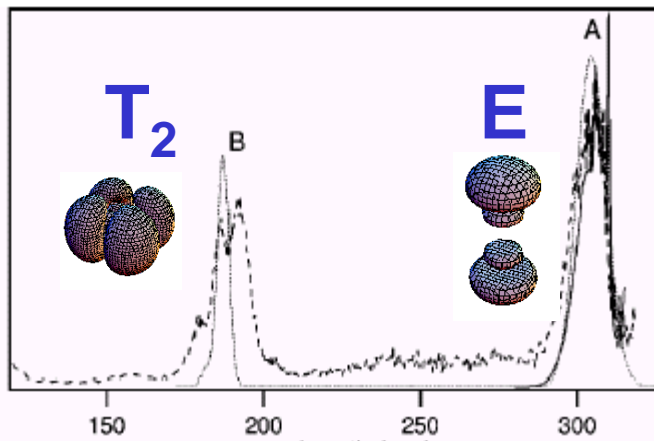
Conduction Band, Free Electrons, Excitons



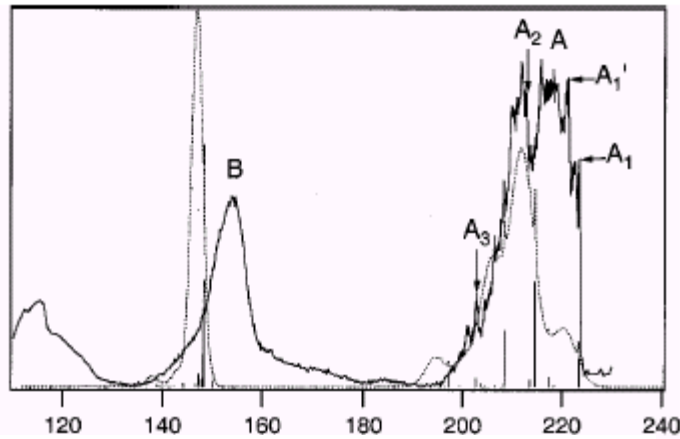
CaF₂ (cubic sites)



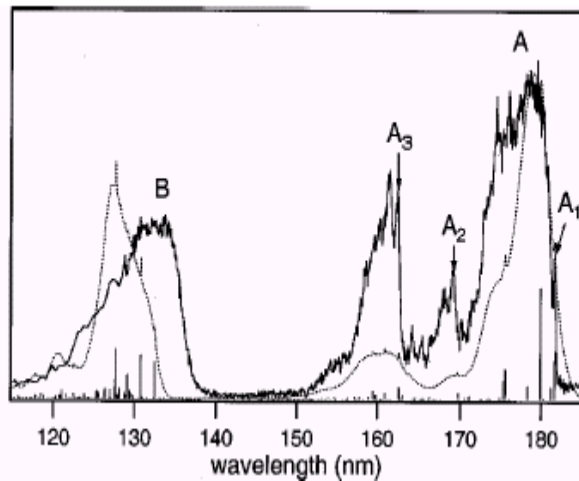
intensity (arb. units)



intensity (arb. units)



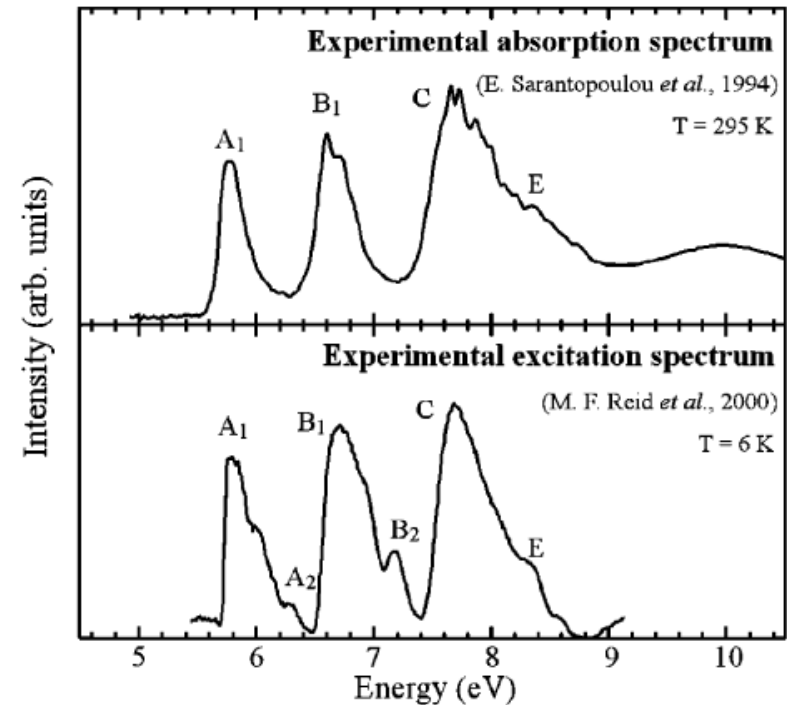
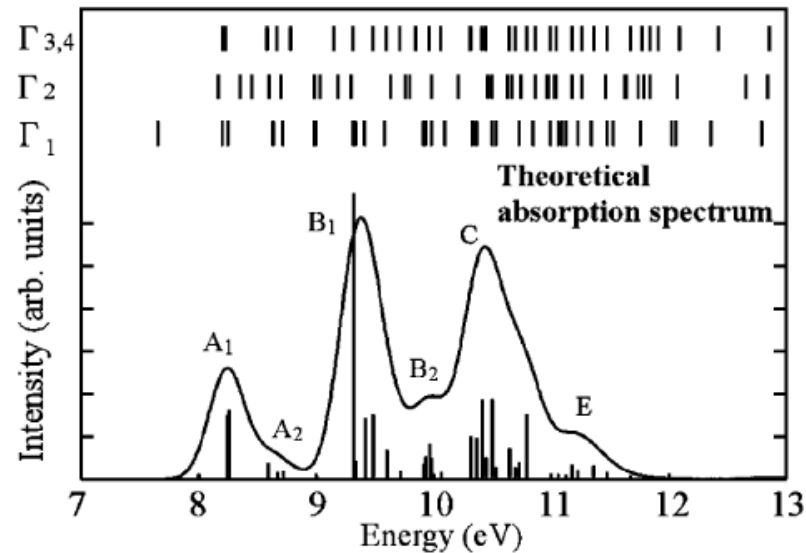
intensity (arb. units)



Energy

First-Principles Calculations

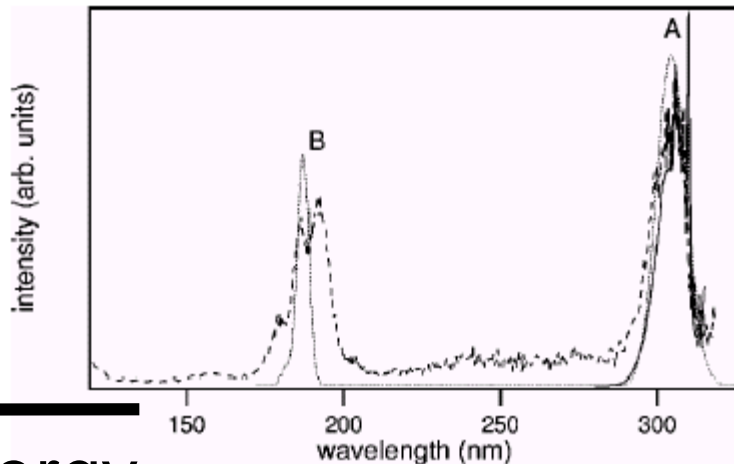
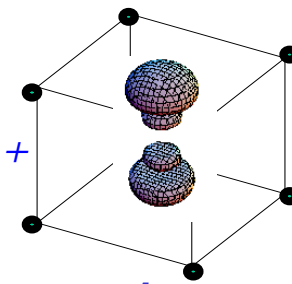
- Relativistic ab-initio calculations are now possible for these systems.
- Ogasawara et al.
J. Solid State Chem. 178, 412 (2005).
 - Calculations for entire series. Some inaccuracies.
- Seijo et al.
J. Chem. Phys. 125, 074511 (2006)
 - Very accurate and detailed calculations for particular ions, including potential surfaces.



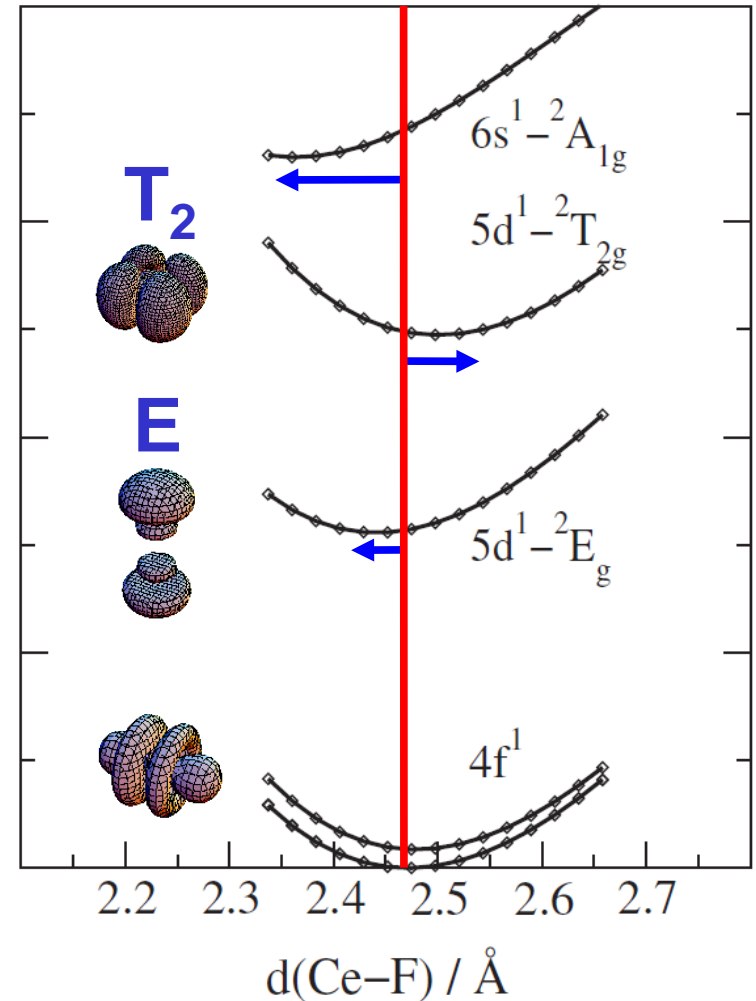
$\text{Pr}^{3+}:\text{LiYF}_4$

Excited-state geometry: $\text{CaF}_2:\text{Ce}^{3+}$

- Pascual, Schamps, Barandiaran, Seijo, PRB 74, 104105 (2006)
 $\text{BaF}_2:\text{Ce}^{3+}$ cubic sites.
- Potential surfaces:
 - 5d E is contracted
 - 5d T_2 is expanded
 - As bond length contracts 6s orbital becomes delocalized.

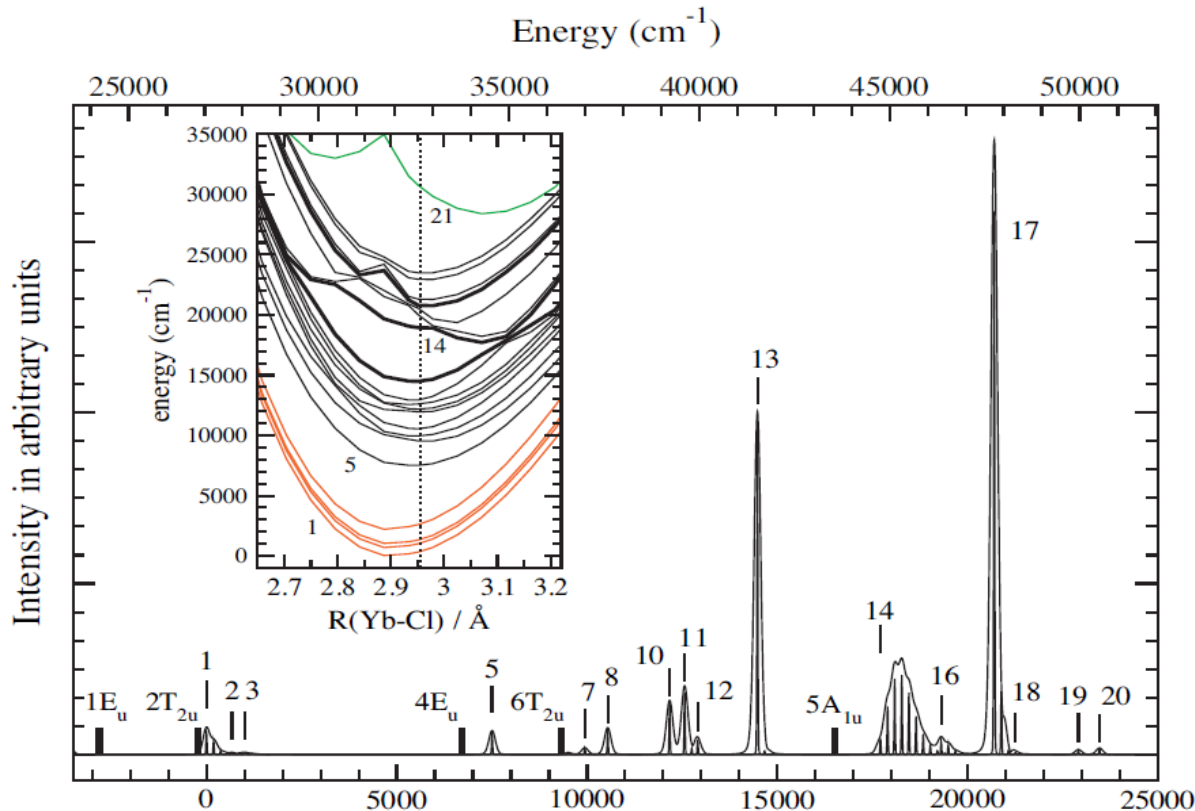


Energy ←

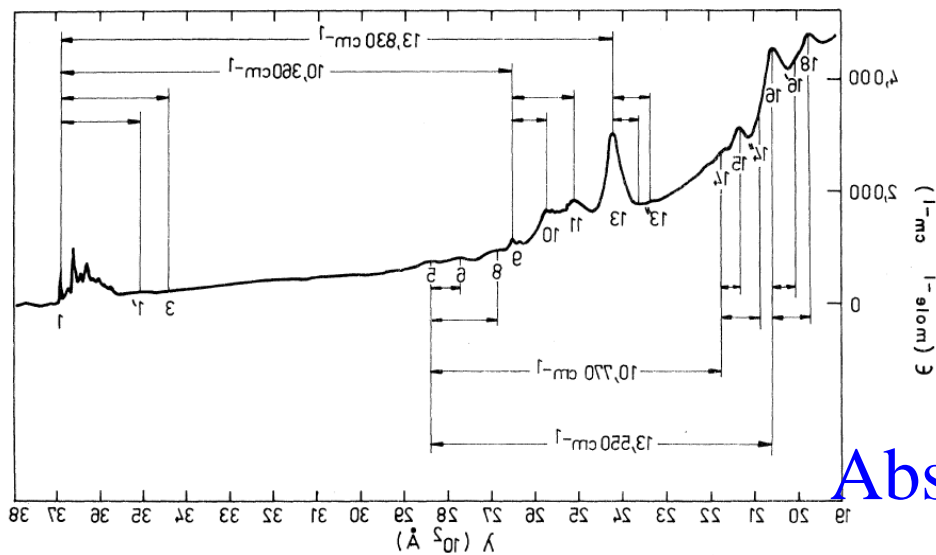


SrCl₂:Yb²⁺

Sánchez-Sanz et al.
 J. Chem. Phys.
 133, 114509 2010



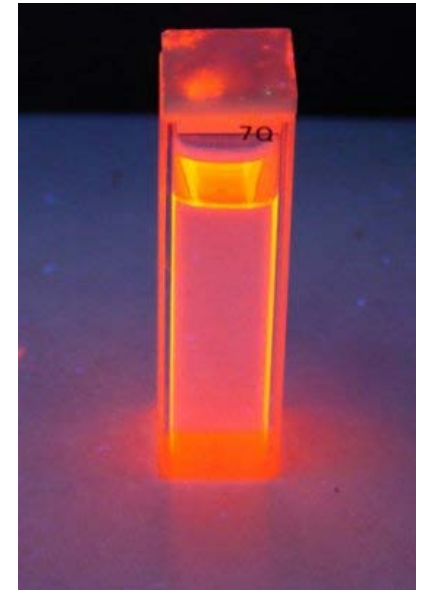
Some of our recent work is on extracting parameters from these calculations.



Absorption

Conclusion

- Effective Hamiltonian for $4f^N$
- Examples of energy level calculations
- Transition intensities
- $4f^{N-1}5d$
- Ab-initio calculations
- Further information and *exercises*:
<http://www2.phys.canterbury.ac.nz/~mfr24/>
<http://www.phys.canterbury.ac.nz/people/reid.shtml>
Email: mike.reid@canterbury.ac.nz



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